

GENERATION OF THIOCARBONYL YLIDES WITH RELEASE OF DISILOXANE
 FROM BIS(TRIMETHYLSILYLMETHYL) SULFOXIDES¹⁾

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Abstract : A novel and general method for generation of thiocarbonyl ylide by release of disiloxane from bis(trimethylsilylmethyl) sulfoxide (**1**) has been found. This method was also demonstrated for generation of aliphatic and aromatic thioaldehyde S-methylide.

Thiocarbonyl ylides have become increasingly attractive for studies on 1,3-dipolar cycloadditions.²⁾ Their formation by thermolysis of 1,3,4-thiadiazole was discovered by Kellogg and his co-workers³⁾ and has been investigated in detail.⁴⁾ Evidence for the existence of thiocarbonyl ylide was provided by Tamagaki and Oae.⁵⁾ We reported a convenient method for generation of trimethylsilylthioaldehyde S-methylide.⁶⁾ However, it is surprising that the parent ylide, thioformaldehyde S-methylide (**3**) has not been reported. We have now succeeded in the generation of ylide (**3**) under the neutral and mild conditions and 1,3-cycloaddition with conjugated dipolarophiles. Our strategy involved the release of disiloxane from bis(trimethylsilylmethyl) sulfoxide (**1a**)⁷⁾ through a pathway related to the sila-Pummerer rearrangement⁸⁾.

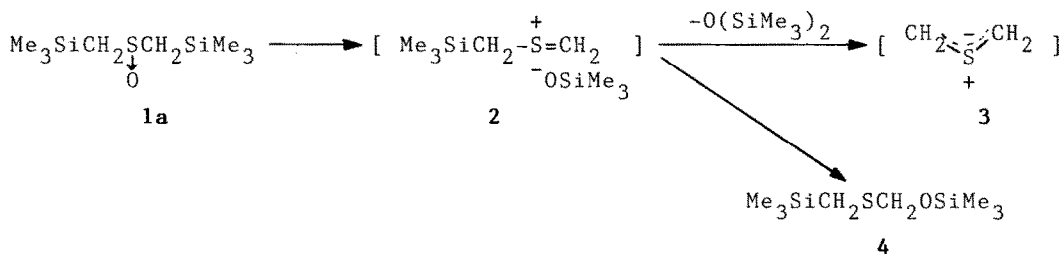
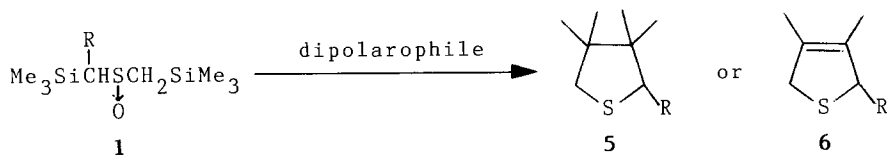


Table I. 1,3-Dipolar Cycloaddition via Thiocarbonyl Ylides



Run	Sulfoxide	Dipolarophile	Cycloadduct	Yield ^a (%)
1	 1a	 $\text{EtO}_2\text{C}-\text{C}(\text{H})=\text{C}-\text{CO}_2\text{Et}$	 5a	80
2	1a	 N-Me	 5b	81
3	1a	 N-Ph	 5c	70
4	1a	$\text{MeO}_2\text{C}-\text{C}\equiv\text{C}-\text{CO}_2\text{Me}$	 6a	61
5	1a	$\text{CH}_2=\text{CHCO}_2\text{Et}$	 5d	55
6	 1b	 $\text{EtO}_2\text{C}-\text{C}(\text{H})=\text{C}-\text{CO}_2\text{Et}$	 5e^b	50
7	1b	 N-Me	 5f^c	65
8	 1c	 N-Me	 5g^d	65

^a Isolated Yield.

^b 2,3-*cis* : 2,3-*trans* = 1 : 1.

^c A single stereoisomer was obtained.; see ref. 12.

^d 2,3-*cis* : 2,3-*trans* = 1 : 5.; see ref. 12.

The required sulfoxide (**1a**) for this reaction was prepared from readily available sulfide by oxidation with m-chloroperbenzoic acid in CH_2Cl_2 at -70°C .⁹⁾ Treatment of 1.5 equiv of **1a** with a dipolarophile in HMPA at 100°C ¹⁰⁾ gave the corresponding cycloadduct. As indicated by the representative example in runs 1-5 (Table I), 1,3-cycloaddition of **1a** proceeded smoothly to give the tetrahydrothiophenes and 2,5-dihydrothiophenes in good to moderate yields.¹¹⁾ This method was also utilized for generation of thiobenzaldehyde S-methylide and phenylthioacetaldehyde S-methylide as shown in runs 6-8.¹²⁾

Treatment of the rearrangement product **4** with a dipolarophile under similar conditions did not give any cycloadduct. This fact suggests that intermediate **2** partitions between the desired ylide **3** and the sila-Pummerer rearrangement product **4**.

Thus we have provided a novel and general method for generation of various thiocarbonyl ylides such as thioformaldehyde S-methylide, and aromatic and aliphatic thioaldehyde S-methylides by employing organosilicon compounds.

References and Notes

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review see: W. P. Weber, "Silicon Reagents for Organic Synthesis" Springer Verlag. Berlin(1983), pp 346-350.

- 9) Sulfoxide **1** was obtained as a sufficiently pure liquid in practice without distillation after the usual workup. **1a**: IR (neat) 1028 cm^{-1} (ν_{SO}). $^1\text{H-NMR}$ (CDCl_3 , 90 MHz) δ 0.22(18H, s), 2.14(2H, d, $J=13.7$ Hz), 2.45(2H, d, $J=13.7$ Hz). **1b**: IR (neat) 1035 cm^{-1} (ν_{SO}). **1c**: IR (neat) 1030 cm^{-1} (ν_{SO}).
- 10) The reaction was completed in 10 min. after addition of a solution of **1a** and a dipolarophile in HMPA to HMPA warmed at 100°C .
- 11) All products gave satisfactory spectral and analytical data consistent with the structures shown. Their $^{13}\text{C-NMR}$ spectra provided conclusive evidence for cycloadducts. $^{13}\text{C-NMR}$ (CDCl_3) δ **5a**: bp $150-151^\circ\text{C}$ (15 mmHg). 14.90(q), 33.33(t), 51.10(t), 61.20(d), 171.80(s). **5b**: mp $90-91^\circ\text{C}$. 22.25(q), 36.00(t), 48.68(d), 177.93(s). **5c**: mp $152-153^\circ\text{C}$. 36.52(t), 48.49(d), 126.45(d), 128.73(d), 129.11(d), 132.51(s), 177.60(s). **5d**: bp $90-91^\circ\text{C}$ (15 mmHg). 14.19(q), 30.66(t), 33.43(t), 33.92(t), 48.33(d), 60.79(t), 169.36(s). **6a**: bp $95-96$ (0.5 mmHg). 39.39(t), 52.45(q), 138.46(s), 164.55(s).
- 12) **5e**: The ratio of both isomers was determined by gas chromatographic analysis. bp $150-152$ (6 mmHg). $^{13}\text{C-NMR}$ (CDCl_3 , 90 MHz) δ 13.76(q), 14.14(q), 33.32(t), 33.43(t), 48.65(d), 51.74(d), 52.12(d), 54.72(d), 56.78(d), 59.17(d), 60.63(t), 60.90(t), 62.12(t), 61.22(t), 127.70(d), 127.98(d), 128.08(d), 128.52(d), 128.73(d), 169.48(s), 171.32(s), 171.86(s). **5f**: mp $166-167^\circ\text{C}$. $^1\text{H-NMR}$ (CDCl_3) δ 2.96(3H, s), 3.15-3.25(1H, m), 3.45-3.60(3H, m), 4.67-4.77(1H, m), 7.30(5H, s). $^{13}\text{C-NMR}$ (CDCl_3) δ 25.08(q), 34.40(t), 50.17(d), 52.23(d), 56.72(d), 128.30(d), 134.69(s), 174.56(s), 177.81(s). Stereochemistry of **5f** is not clear. **5g**: 2,3-*cis* isomer mp $72-72^\circ\text{C}$. $^1\text{H-NMR}$ (CDCl_3) δ 2.70-2.92(2H, m), 3.04(3H, s), 3.20-3.76(5H, m), 7.25(5H, s). $^{13}\text{C-NMR}$ (CDCl_3) δ 25.14(q), 34.73(t), 35.76(t), 49.14(d), 49.74(d), 54.99(d), 126.72(d), 128.51(d), 128.62(d), 139.78(s), 177.98(s). 2,3-*trans* isomer mp $84-85^\circ\text{C}$. $^1\text{H-NMR}$ (CDCl_3) δ 2.92(2H, d, $J=7.2$ Hz), 2.96(3H, s), 3.12-3.56(4H, m), 3.92(1H, td, $J=7.2$ Hz, 1 Hz), 7.28(5H, s). $^{13}\text{C-NMR}$ (CDCl_3) δ 25.30(q), 33.86(t), 41.61(t), 48.65(d), 52.71(d), 53.69(d), 127.00(d), 128.62(d), 128.63(d), 129.17(d), 138.51(s), 139.78(s), 177.98(s).

(Received in Japan 14 June 1986)